## A SIMPLE ENANTIOCOMPLEMENTARY ROUTE TO PROSTANOIDS;

INVERSION OF CHIRALITY OF 2,5-DIHYDROXY-CYCLOPENT-
-3-ENYL-ACETIC ACID LACTONE DERIVATIVES
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Abstract: An enantiocomplementary route to $P G A_{2}$ and $P G E E_{2}$ synthons is presented.

The enantiomers of unsaturated bicyclic lactone ${ }^{1} \underline{1}$ are valuable chiral intermediates in the Corey ${ }^{2}$ and Stork ${ }^{3}$ synthesis of $\mathrm{PGA}_{2}$ and $\mathrm{PGE}_{2}$, respectively.



The precursor (-) $\underline{1 b}$ was reported ${ }^{2}$ to be readily available from (-) $\underline{2}$ without formation of appreciable by-products by a sequence outlined below:


We examined the effect of pH upon the products of iodolactonisation, anticipating the interference of $O H$ group in 4 a with the favorably oriented lactone moiety, that might induce facile $0-0$ acyl migration ${ }^{4}$, which leads to the formation of 5 a. Iodolactonisation ${ }^{5}$ conducted at various pH values ( $5 \leqq \mathrm{pH} \leqq 9$ ) with $1.1-1.2$ eq iodine actually reveals striking dependence of product distribution on the pH value as reflected by ${ }^{l_{H}}$ NMR spectra of the crude products. With TLC, however, only one new spot could be detected in various eluent mixtures. Further intervention of acyl migration (translactonisation) could be excluded by blocking the pertinent of group. Both acetylation $\left(1.2\right.$ eq $\mathrm{Ac}_{2} \mathrm{O}$, 1.2 eq pyridine, 0.2 eq 4 -dimethyl-aminopyridine in $\mathrm{CH}_{2} \mathrm{Cl}_{2}, 20{ }^{\circ} \mathrm{C}, 1 \mathrm{~h}$ ) and tetrahydropyranylation (2 eq DHP, cat. pyridinium tosylate in $\mathrm{CII}_{2} \mathrm{Cl}_{2}, 20^{\circ} \mathrm{C}, 4 \mathrm{~h}$ ) yielded a chromatographically inseparable product mixture, however, t-butyldimethylsilyl (4b and 5 b ) ${ }^{6,8}$ or 2 -methoxy-2-propyl ( $\underline{4 c}$ and $\underline{5 c})^{7,8}$ derivatives were easily detected on $T L C\left[R_{f} \underline{4 b}: 0.38\right.$, 5b: 0.27 (hexane-ethyl acetate 3:1), 4c: 0.50 , 5c: 0.33 (hexane-ethyl acetate $\left.\left.2: 1+1 \% \mathrm{NEt}_{3}\right)\right]$ and could be readily separated by column chromatography.


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a, $R=H$
b, $R=t-$ BuMe $_{2} \mathrm{Si}-$
c, $R=\mathrm{Me}_{2}(\mathrm{MeO}) \mathrm{C}-$
$\mathrm{d}, \mathrm{R}=\mathrm{Ac}$
e, $R=T H P$




In iodolactonisations, performed at $\mathrm{pH}=5-6$ followed by protection and separation of isomers, 4 b or 4 c was formed in an amount four times as high as 5 b or 5 c whereas in experiments conducted at $\mathrm{pH}=8$ the inverse isomeric ratio ( $\underline{4}: \underline{5}=1: 4$ ) was obtained. It has been assumed that the ratio $\underline{4}$ : 5 formed in aqueous solution by trans-lactonisation remains essentially unchanged during protection, as the same $4 \underline{b}: \underline{5 b}$ and $\underline{4 c}: 5 c$ ratios were obtained irrespective of the character of the catalysts applied (imidazole or pyridinium tosylate).

Dehydroiodination of 4 b or 4 c with 1.5 eq DBU in dry THF ( $20^{\circ} \mathrm{C}, 4-8 \mathrm{~h}$ ) gave optically pure ( - ) 1 b or 1 c in contrast to 5 b or 5 c , which, upon the same treatment yielded $(+) \underline{l b}$ or 1 c in identical magnitude but opposite sign of rotation in 90-95 \% yield.

Details of several experiments performed at $\mathrm{pH}=5.5$ and 8 are summarized in the Table.

| start.mat ${ }^{\text {², }}$ |  | 4 |  | 1 |  | 5 |  | 1 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\underline{2}[a$ |  | $\overbrace{\mathrm{p}}^{\mathrm{h}}$ | ${ }^{[\alpha]} \mathrm{D}$ |  | $[\alpha]_{D}$ | $\begin{aligned} & \mathrm{o}_{\mathrm{o}} \mathrm{p}^{-} \end{aligned}$ | ${ }^{[a]}$ D | $\stackrel{3}{ }{ }^{\prime}$ | ${ }^{[\alpha]} \mathrm{D}$ |
|  |  | $\left(\mathrm{R}=\mathrm{t}-\mathrm{BuMe}_{2} \mathrm{Si}-\right)^{\text {C }}$ |  |  |  |  |  |  |  |
| -106 | 5.5 | $\begin{aligned} & 81.6 \\ & 84.0 \end{aligned}$ | +28.2 | 54.0 | -27.4 | $\begin{aligned} & 18.4 \\ & 16.0 \end{aligned}$ | +51.2 | 10.0 | +26.8 |
| +106 | 8.0 | 23.7 | -28.9 | 10.3 | +26.9 | 76.3 | -50.8 | 52.5 | -27.2 |
|  |  | $\left(\mathrm{R}=\mathrm{Me}_{2}(\mathrm{MeO}) \mathrm{C}-\right)^{\mathrm{m}}$ |  |  |  |  |  |  |  |
| -106 | 5.5 | $7 \overline{6} .0$ | -17.1 | 52.0 | -100. 3 | $24.0$ | +82.3 | 9.7 | +99.6 |
| +106 | 8.0 | $\begin{aligned} & 19.0 \\ & 17.5 \end{aligned}$ | +17.9 | 10.9 | +100.4 | $\begin{aligned} & 81.0 \\ & 82.5 \end{aligned}$ | -81.9 | 53.5 | -99.5 |

$h$ : determined by HPLC (upper value): $p$ : determined by prep. oolumn chrom; $y$ : isolatod overall yield; $[\alpha]_{D}$ values refer to $\mathrm{CHCl}_{3}(c)$ or to $\mathrm{MeOH}(\mathrm{m})$, ( conc=1 \%).

Graphic illustration of the enantiocomplementarity is given below.


Heavy and light arrows represent respective processes of 52-54 \% and 10-12 \% overall yields, including separation during the process; pH data refer to conditions used in iodolactonisation.

Undesirable isomers can be deprotected and recycled aftcr scparation instead of being converted to 1 (light arrows), thus allowing further increase in effectivity.
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5. In a typical procedure $10.0 \mathrm{~g} \underline{2}$ was dissolved in 95 ml in NaOH and after stirring for 3 hrs the pH was adjusted to 8 or 5.5 by adding solid $\mathrm{CO}_{2}$ or In $\mathrm{NaHSO}_{4}$, respectively, $22.8 \mathrm{~g} \mathrm{I}_{2}$ was then added in one portion and the reaction mixture was stirred overnight. After extraction with ethyl acetate the excess iodine was destroyed with $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}$, the organic phase was dried and evaporated to give $4 \underline{a}+5 \mathrm{a}, 19 \mathrm{~g}$ (94\%). This mixture was treated either with 1.5 eq t-butyl-dimethylsilyl chloride ${ }^{9}$ and 2 eq imidazole in DMF (rt, 20 hrs ) to give $\underline{4 \mathrm{~b}}+\underline{5 \mathrm{~b}}$ or with 2 eq 2 -methoxy-1-propene and cat. amount of pyridinium tosylate in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ (rt, 6 hrs ) to give $\underline{4 \mathrm{c}}+\underline{5 \mathrm{c}}$ after quenching with triethylamine and evaporation of the solvent. Purification and separation of the isomers were performed by chromatography on silica gel using hexane-ethyl acetate $3: 1(\underline{4 b}+\underline{5 b})$ or hexane-ethyl acetate $2: 1$ containing $0.5 \% \mathrm{NEt}_{3}(\underline{4 c}+5 \mathrm{C})$ as eluent.
 $2.7-2.3(4 \mathrm{H}, \mathrm{m}), 0.9 \mathrm{O}(9 \mathrm{H}, \mathrm{s}), 0.14(6 \mathrm{H}, \mathrm{d})$
4b: $5.15(1 \mathrm{H}, \mathrm{dd}), 4.60(1 \mathrm{H}, \mathrm{m}), 4.35(1 \mathrm{H}, \mathrm{m}), 3.2-2.1(5 \mathrm{H}, \mathrm{m}), 0.90(9 \mathrm{H}, \mathrm{s})$, 0.08(6H,s)

1b: $6.0(2 \mathrm{H}, \mathrm{s}), 5.26(1 \mathrm{H}, \mathrm{d}), 4.87(1 \mathrm{H}, \mathrm{d}), 3.45-2.25(3 \mathrm{H}, \mathrm{m}), 0.92(9 \mathrm{H}, \mathrm{s})$, $0.10(6 \mathrm{H}, \mathrm{s})$
7. ${ }^{1} \mathrm{H}$ NMR $\delta$ in $\mathrm{ppm}\left(\mathrm{CDCl}_{3}\right)$ 5c: $4.98(1 \mathrm{H}, \mathrm{m}), 4.39(1 \mathrm{H}, \mathrm{dd}), 4.14(1 \mathrm{H}, \mathrm{m}), 3.29(3 \mathrm{H}, \mathrm{s})$, 1.33, $1.39(3 \mathrm{H}, \mathrm{s})$
$4 \mathrm{C}: ~ 5.13(1 \mathrm{H}, \mathrm{d}), 4.77(1 \mathrm{H}, \mathrm{m}), 4.36(1 \mathrm{H}, \mathrm{dd}), 3.18(3 \mathrm{H}, \mathrm{s}), 1.34,1.31(3 \mathrm{H}, \mathrm{s})$
8. Satisfaytory combustion data, $I R$ and ${ }^{1} H_{N M R}{ }^{6,7}$ spectra were obtained for all the new compounds.
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